

# Microfluidic devices for cell culturing and electrochemical sensing of hydrogen peroxide and nitrite

Daniel Rojas<sup>1,2</sup>, Juan F. Hernández-Rodríguez<sup>1</sup>, Flavio della Pelle<sup>2</sup>, Michele Del Carlo<sup>2</sup>, Dario Compagnone<sup>2</sup>, Alberto Escarpa<sup>1,3</sup>

<sup>1</sup>Department of Analytical Chemistry, Physical Chemistry and Chemical Engineering, Faculty of Sciences University of Alcalá, E-28871 Alcalá de Henares, Madrid, Spain.

<sup>2</sup>Faculty of Bioscience and Technology for Food, Agriculture and Environment University of Teramo 64023, Teramo (Italy).

<sup>3</sup>Chemical Research Institute "Andres M. del Rio", University of Alcalá, E-28871, Madrid, Spain

## Abstract

Integration of electrochemical detection (ED) in cell culture is an advantageous strategy to detect the secretions from cultured cells. This is especially interesting for unstable species since they are not diluted prior to detection, so the minute quantities of species secreted be detected and quantified. Reactive Oxygen Species (ROS) and Reactive Nitrogen Species (RNS) are a group of compounds of special interest because of their critical role in physiological processes like cellular signalling and immunological activity. However, an overproduction may cause the so-called oxidative stress (OS) which is able to cause damage to lipids, proteins or DNA. These alterations promote pathophysiological conditions such as diabetes, cancer, Alzheimer's and Parkinson's disease. To study this phenomenon, Hydrogen peroxide ( $H_2O_2$ ) and nitrite ( $NO_2^-$ ) were selected as indicators of ROS and RNS respectively for their stability and for being the stable end-products for each group. In this work, the ED of these compounds has been studied employing different electrode materials like Pt, carbon black (CB) and Prussian Blue. The microfluidic system coupled with ED was further employed to culture different cell lines and to follow their response in terms of  $H_2O_2$  and  $NO_2^-$  concentrations released after treatment with calcimycin.

## Introduction

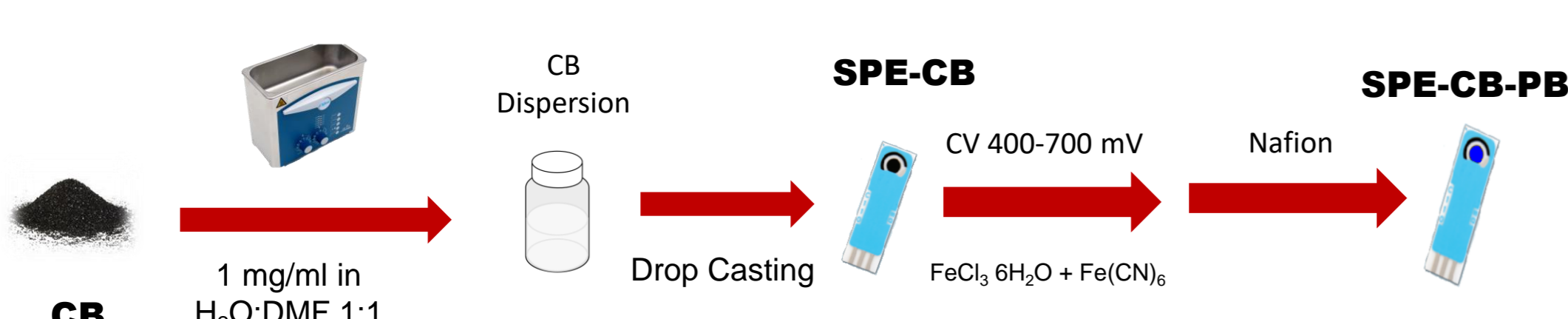
Oxidative Stress is defined as an imbalance between oxidant stressors and antioxidant defences, this physiological status leads to several diseases such as cancer, ischemia, atherosclerosis, Alzheimer's and Parkinson's disease (PD). Therefore, there is a growing interest in the quantification reactive oxygen species (ROS) and reactive nitrogen species (RNS). Hydrogen peroxide and nitrite are commonly used as indicators of oxidative stress due to its relative stability in contrast to superoxide, nitric oxide or peroxynitrite. Different analytical strategies have been proposed for their detection such as chemiluminescence, fluorescence, and electrochemical techniques. Among these, electrochemical sensors are very appealing for their simplicity, speed, sensitivity, miniaturization and cost-effectiveness. Microfluidic systems allow the real-time measurement of this response on modified PB electrodes. Therefore, the combination of electrochemical detection with microfluidic devices are very appealing for fast and decentralized analysis in cell cultures. As cellular model, macrophages are commonly chosen as a cell line for oxidative stress studies because they are a natural barrier against infection and inflammation. In order to neutralize their targets they usually produce and secrete powerful radicals to the extracellular medium which facilitate the study of their response towards oxidative stressors.

## Objectives

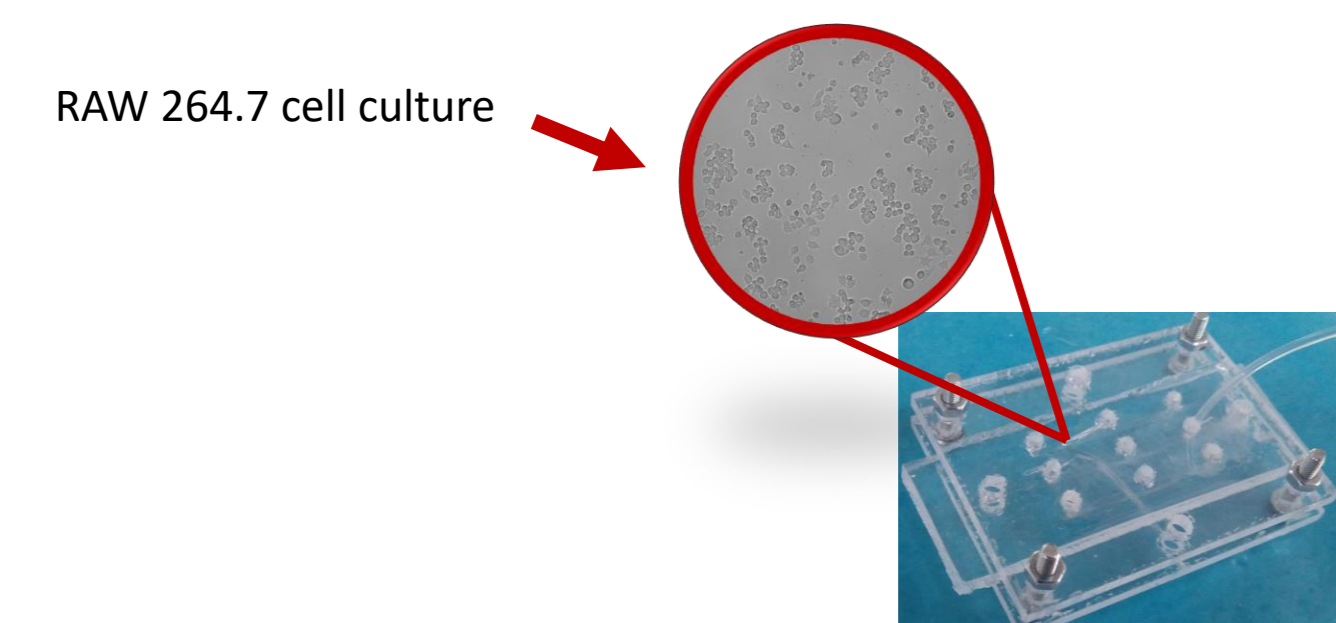
- To develop a new microfluidic platform able to perform electrochemical on-line measurements in cell cultures.
- To test the developed platform towards the detection of  $H_2O_2$  and  $NO_2^-$  released by living cell populations in a integrated platform.

## Experimental

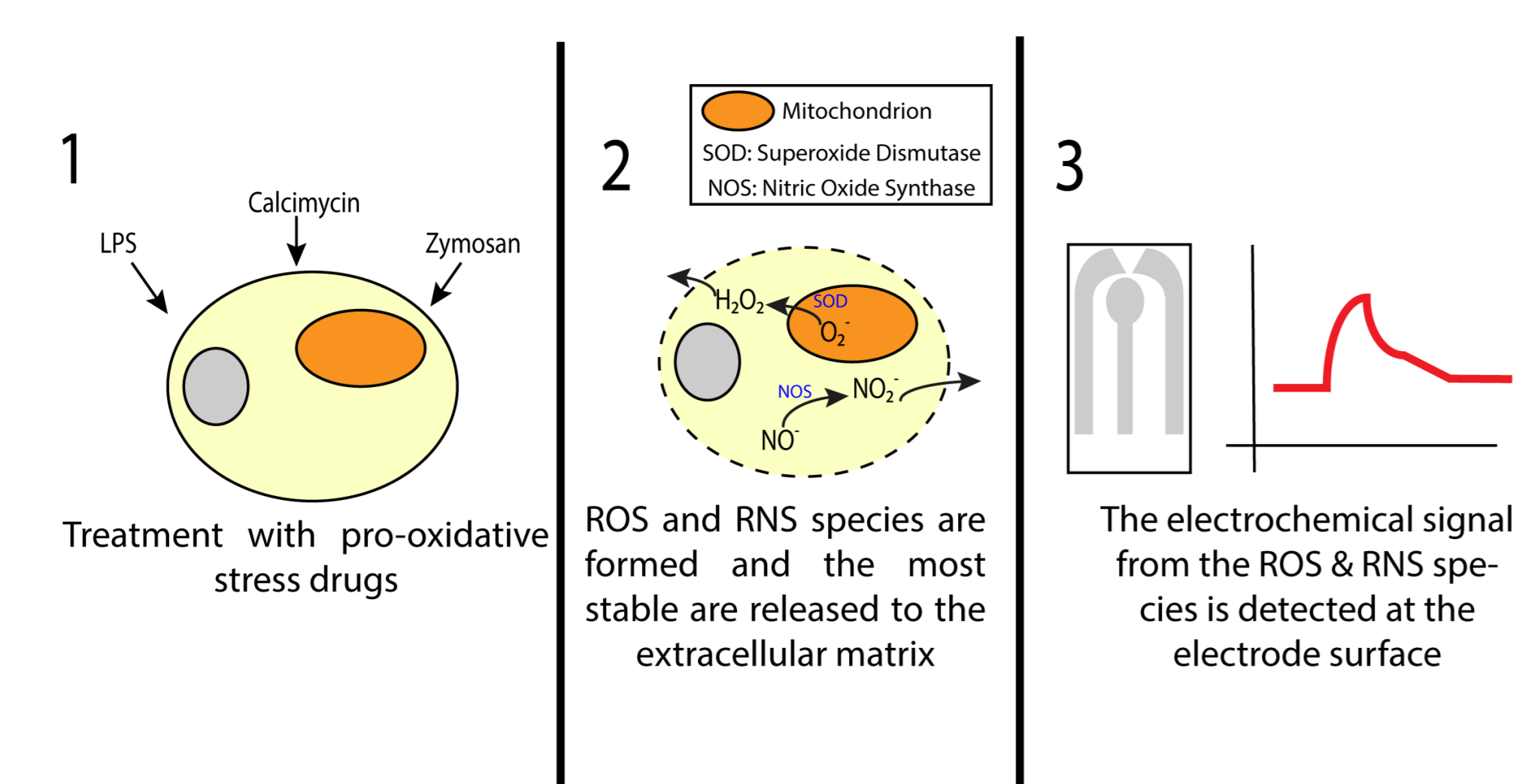
### CB-PB electrodes preparation



### Microfluidic device

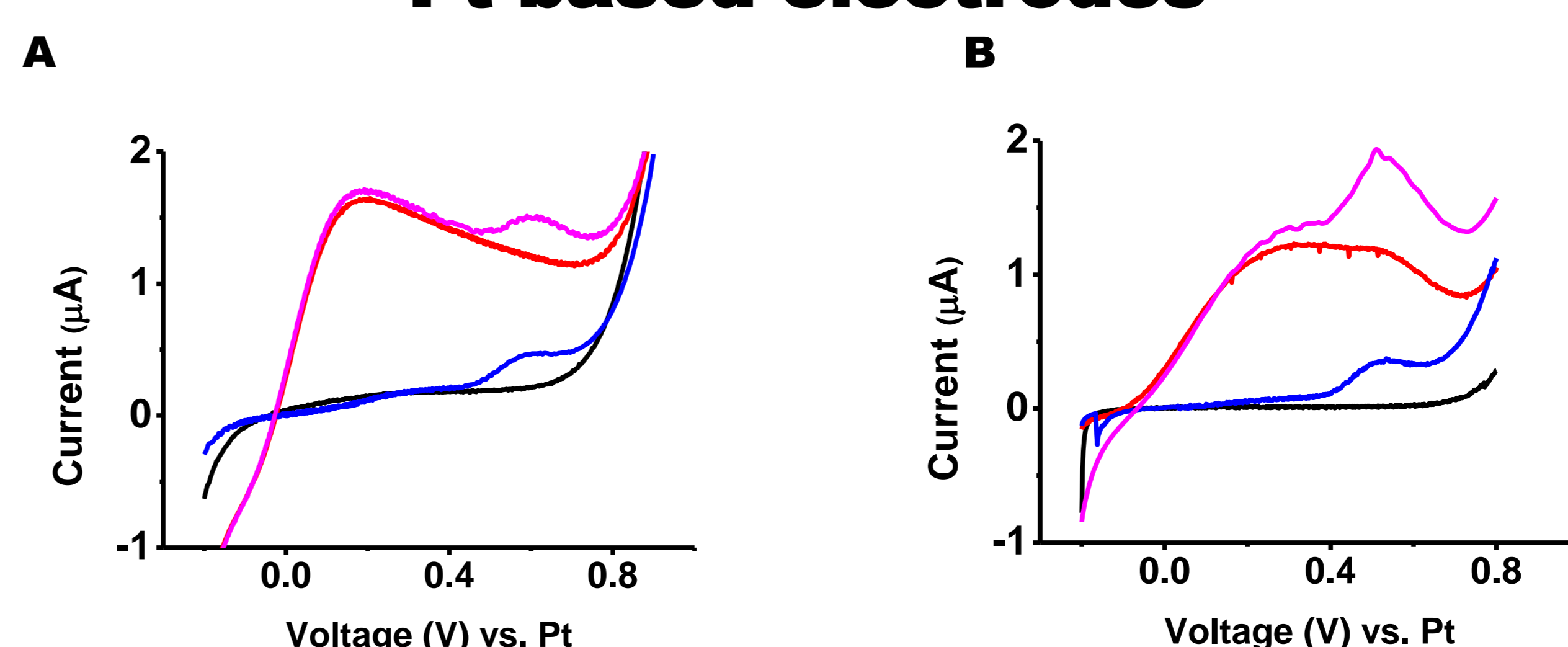


### Assay workflow



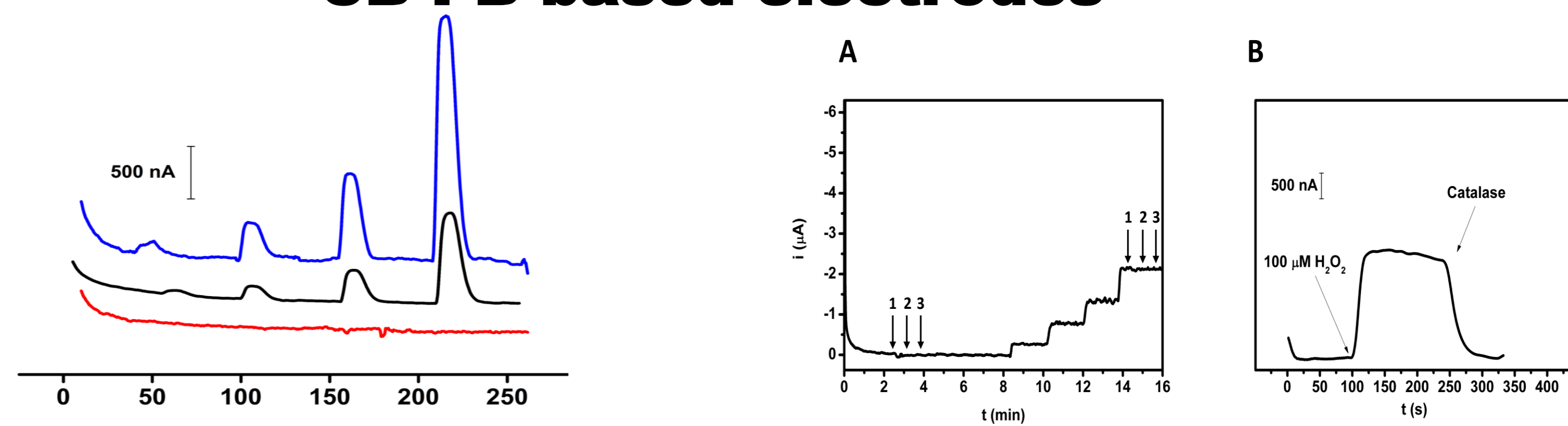
## Results

### Pt based electrodes



A) LSV of PBS (black),  $H_2O_2$  (red),  $NO_2^-$  (blue) and an equimolar mixture (magenta) at 1 mM concentration B) HDV electrode PBS (black),  $H_2O_2$  (red),  $NO_2^-$  (blue) and an equimolar mixture (magenta) at 1 mM concentration. Flow rate  $50 \mu L \text{ min}^{-1}$

### CB-PB based electrodes

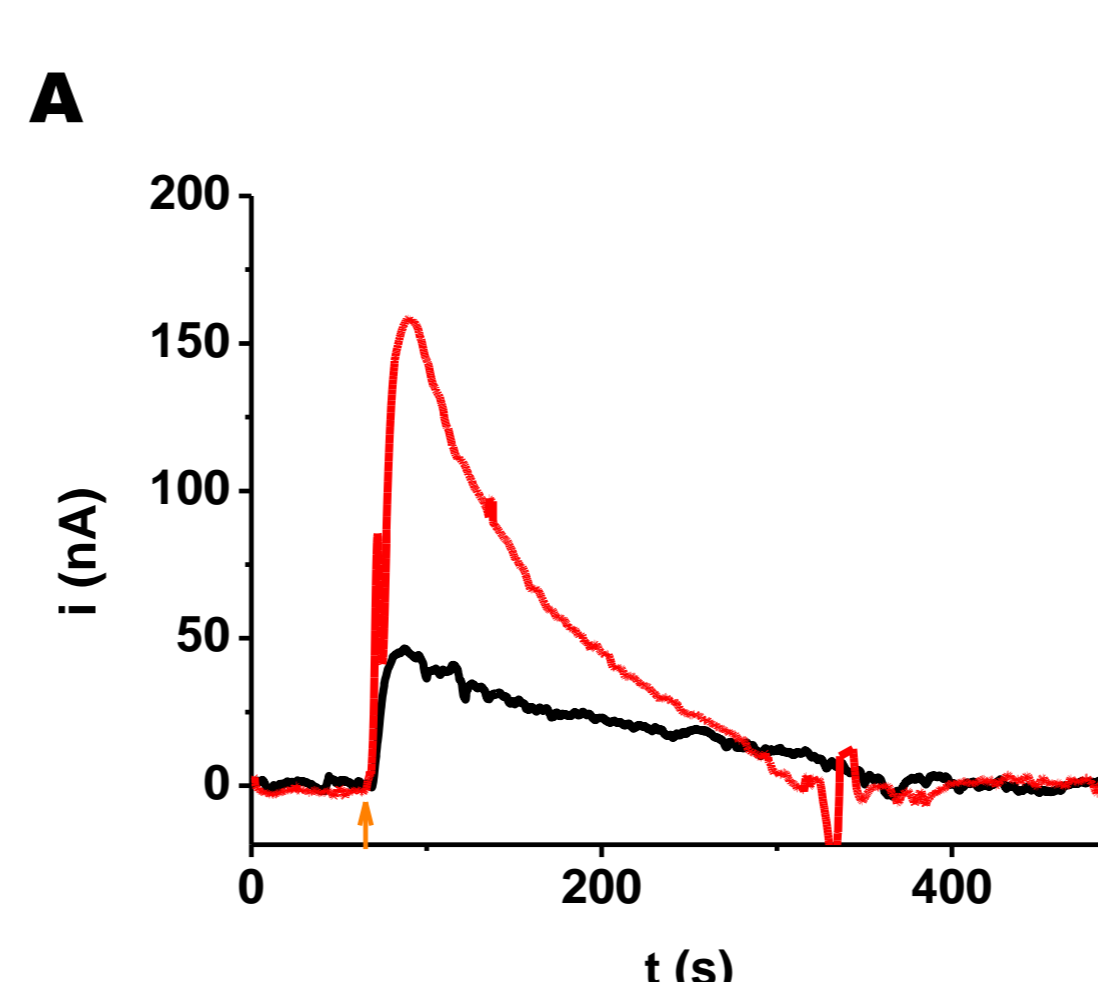


A) Amperometry signals in FIA for 5, 10, 20 and 50  $\mu M$  of  $H_2O_2$  in Phosphate Buffer (pH=7.4) SPE-CB (red line), SPE-PB (black line) and SPE-CB/PB (blue line).

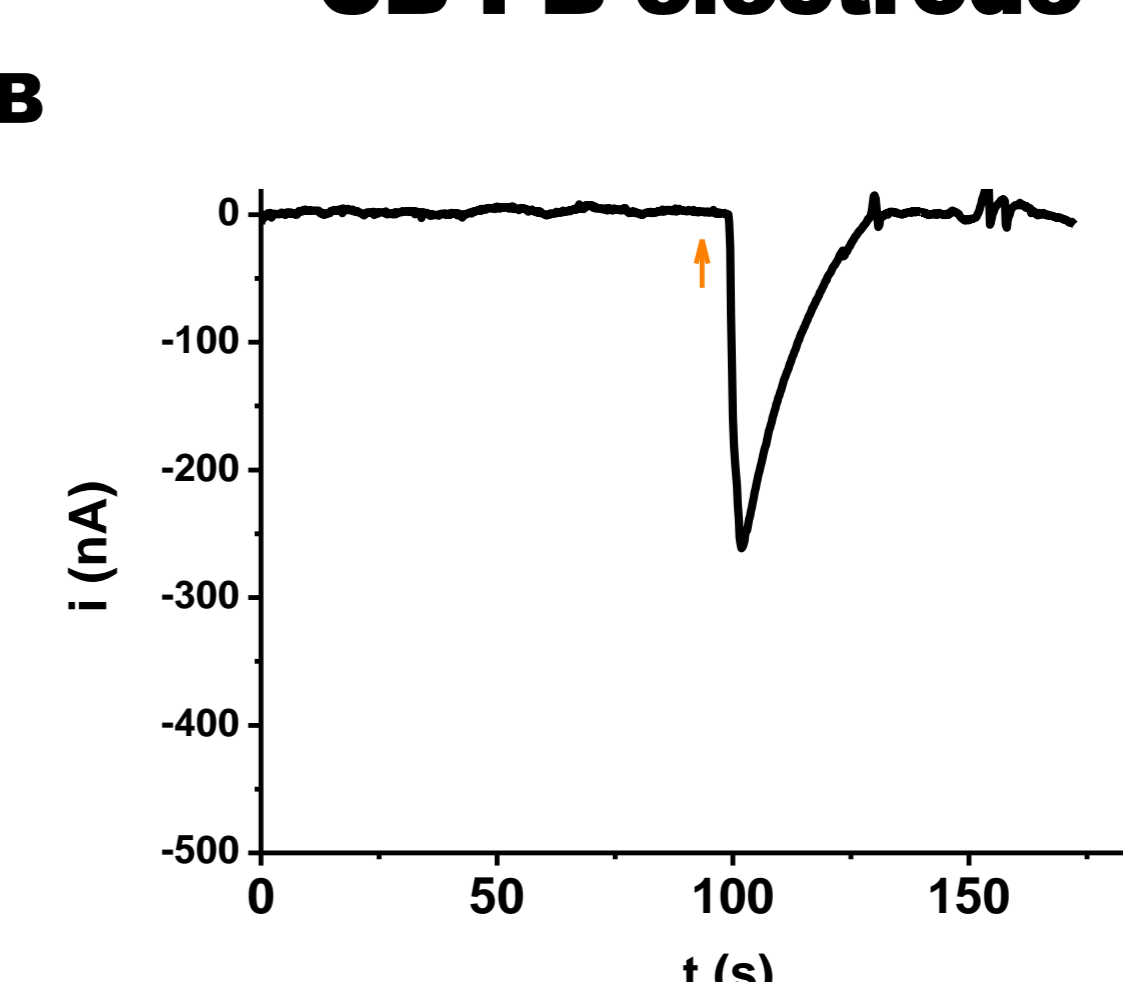
B) Selectivity of the electrode towards 100  $\mu M$  of  $H_2O_2$  spiked in the cell culture without cells. E=50 mV vs Ag

### RAW 264.7 cells responses to calcimycin

#### Pt-based electrode



#### CB-PB electrode



Responses of RAW 264.7 cell population towards 5  $\mu M$  of calcimycin recorded with A) Pt electrodes at 0.3 (black) and 0.85 (red) V vs Ag B) CB-PB electrodes at -0.05 V vs Ag

## Conclusions

- Secretions of Raw 264.7 cells produced by calcimycin were successfully detected using two different electrodes.
- Pt-based electrode is able to measure the end products of ROS ( $H_2O_2$ ) and RNS ( $NO_2^-$ )
- PB-based electrode allowed the interference free detection of hydrogen peroxide.
- These results pave the way for an optimization of the microfluidic device to perform (bio)chemical assays in different cellular conditions.

## References

- Rojas, D.; Della Pelle, F.; Del Carlo, M.; D'Angelo, M.; Dominguez-Benot, R.; Cimini, A.; Escarpa, A.; Compagnone, D. Electrodeposited Prussian Blue on Carbon Black Modified Disposable Electrodes for Direct Enzyme-Free  $H_2O_2$  Sensing in a Parkinson's Disease in Vitro Model. *Sensors Actuators, B Chem.* **2018**, *275*, 402–408.
- Liu, H.; Weng, L.; Yang, C. A Review on Nanomaterial-Based Electrochemical Sensors for  $H_2O_2$ ,  $H_2S$  and  $NO$  inside Cells or Released by Cells. *Microchim. Acta* **2017**, *184* (5), 1267–1283.

